

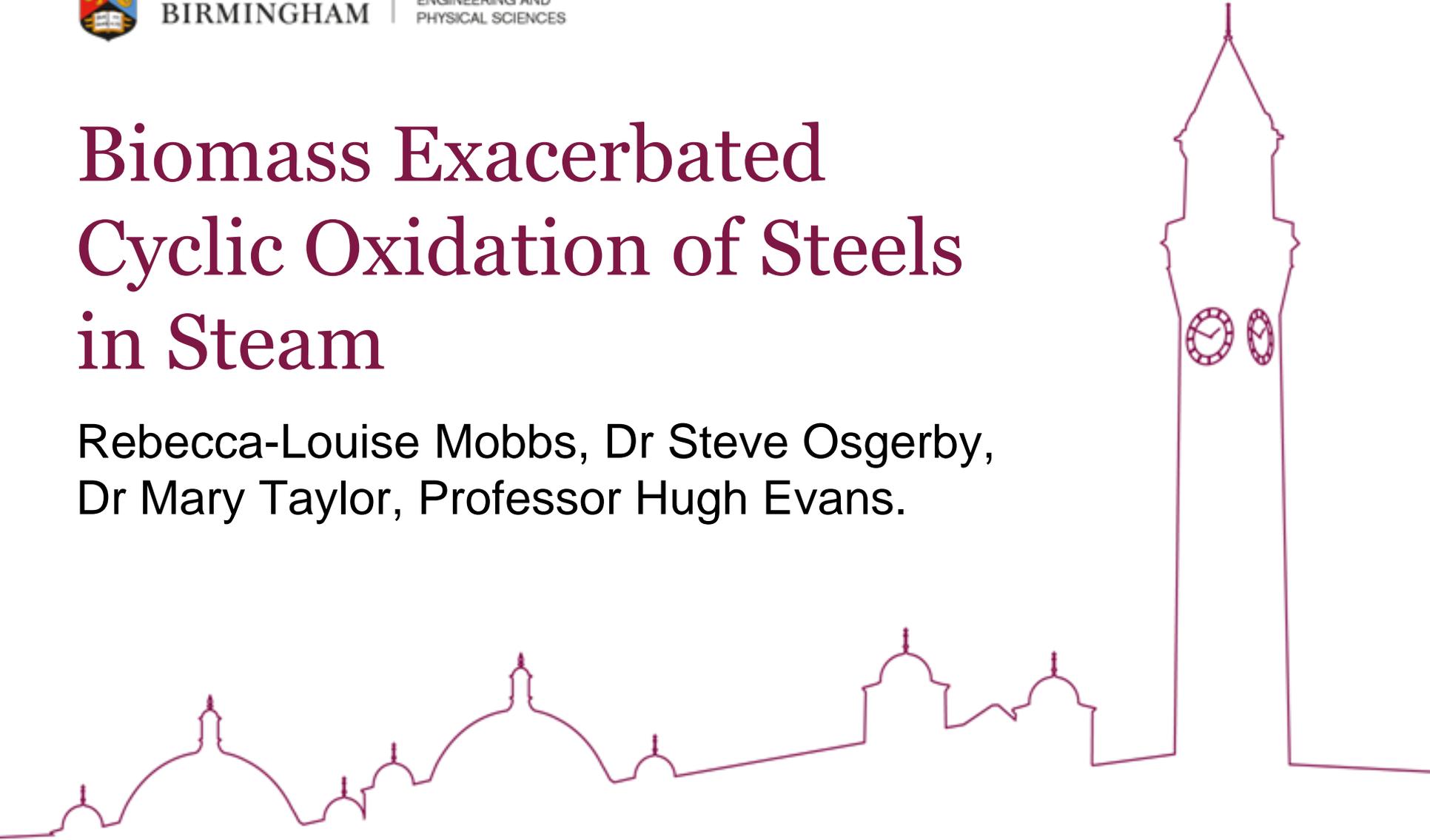


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Biomass Exacerbated Cyclic Oxidation of Steels in Steam

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Introduction

- Alloys undergo oxidation at high temperature.
 - Multilayered oxides of different compositions form.
 - Modification of the underlying alloy results.
- Rate of oxidation is exacerbated under steam conditions.
- On cooling stresses develop in the oxide layer resulting in loss of the oxide (spallation).
- On reheating oxides continue to reform.
- Load following biomass plants will lead to greater thermal cycling.



Introduction

- Spallation leads to:
 - Damage of components down stream,
 - Blockage of tubes,
 - Tube ruptures,
- Unscheduled / more frequent maintenance shutdowns.



Questions

1. How will the cyclic nature of biomass power plants affect the oxidation behaviour of the heat exchanger tubing?
2. When does spallation become a big enough problem that maintenance is required?



Aims and Objectives

- ❑ **To investigate the cyclic steam oxidation and spallation behaviour of an austenitic stainless steel currently used as heat exchanger tubing.**
- ❑ To simulate the effect of the oxidation processes on the degradation of the alloy.



Materials

- Austenitic stainless steel TP347H FG.
 - Grain size = 20 μm
- Composition (wt.%):

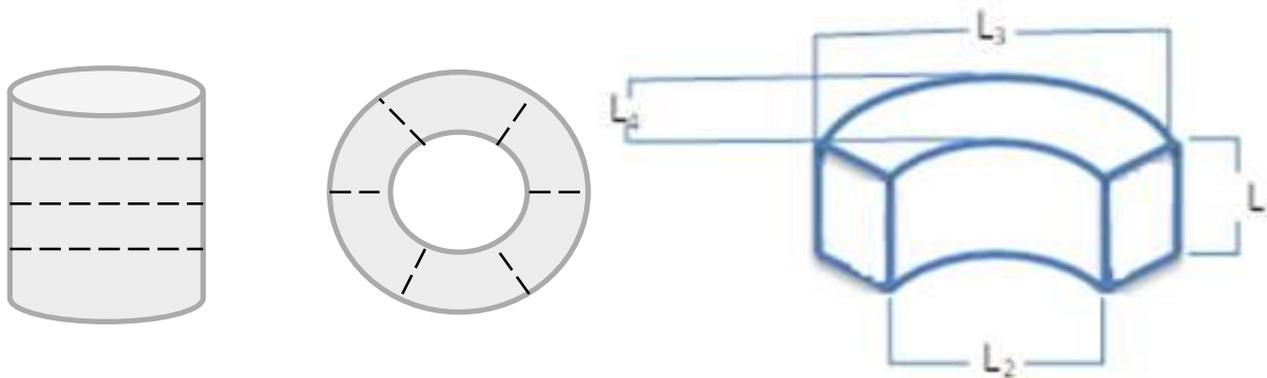
C	Si	Mn	S	P	Ni	Cr	Nb	Fe
0.09	0.4	1.48	0.001	0.026	11.34	18.21	0.88	bal

- Commonly used as superheater and reheater tubing in fossil fired power plants.



Sample Preparation

- Samples are sectioned from standard pipe as installed in plant.

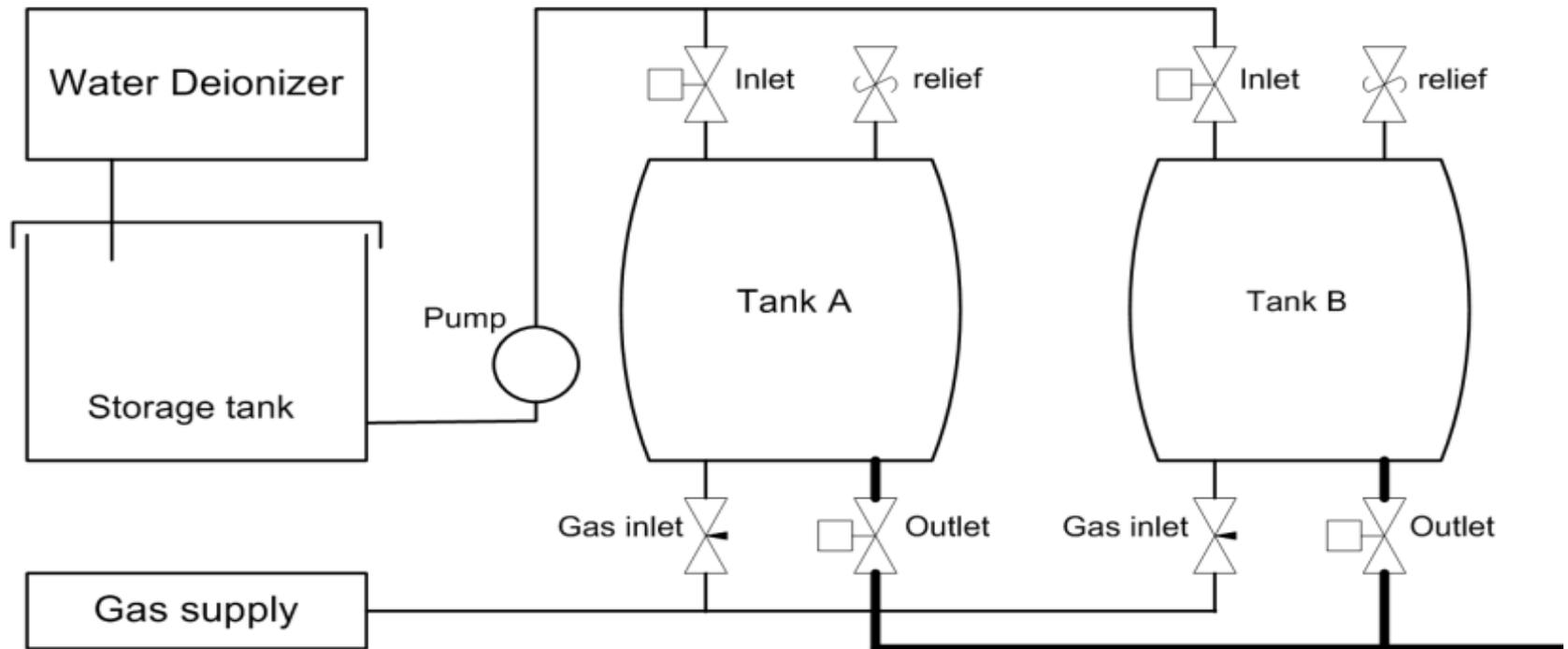


- $L_1 = 10$ mm; $L_2 = 14$ mm; $L_3 = 18$ mm; $L_4 = 4.5$ mm
- Inner surface is of interest – steam side of power plant.

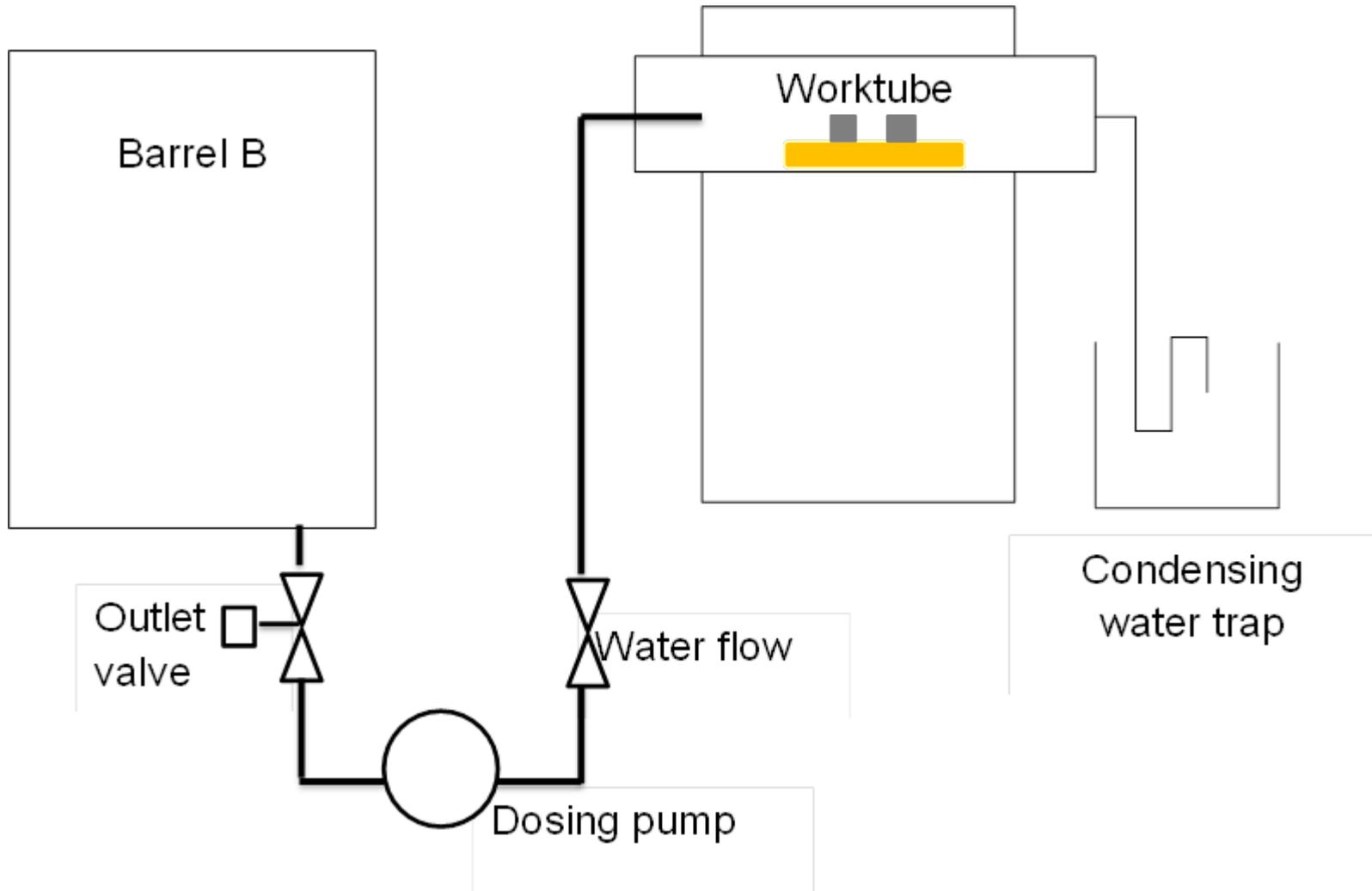


Experimental Work

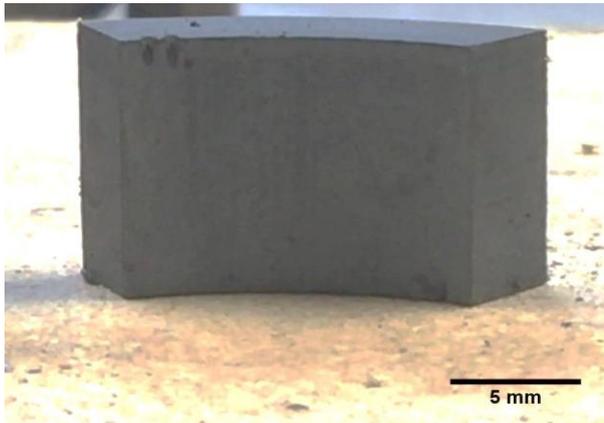
- Exposure to steam at 923 K for 50, 100, 300, 500, 750 and 1000 hours.
- Atmospheric pressure.



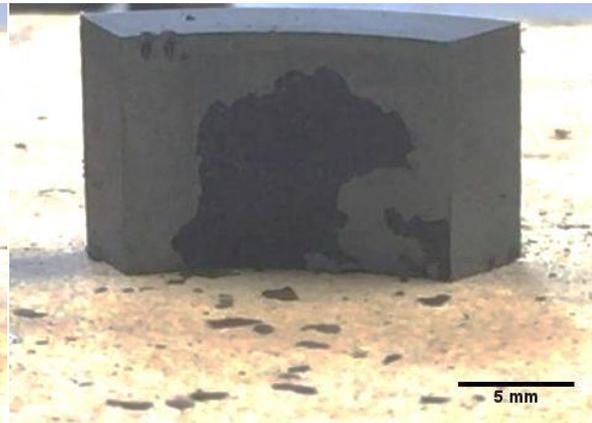
Experimental Work



Spallation



Before spallation.
At temperature.



During spallation.
On cooling.

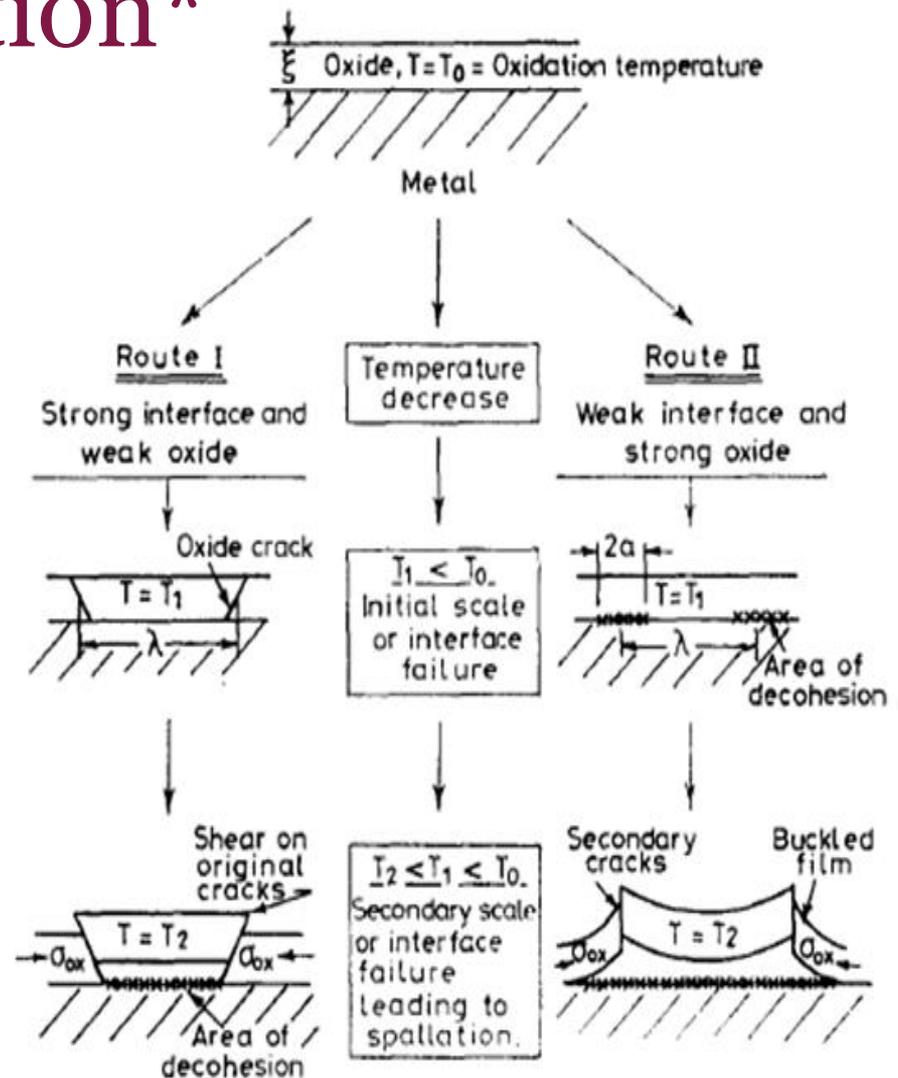


After spallation.
Room
temperature.



Routes to Spallation*

- Two things are necessary to initiate spallation:
 - Decohesion along the spallation interface (delamination)
 - Cracks through the oxide layer to the interface.



Routes to Spallation*

□ Route I: Wedging.

$$\Delta T_W = \sqrt{\frac{\gamma_F}{\xi E_{ox} (\alpha_m - \alpha_{ox})^2 (1 - \nu_{ox})}}$$

$$\gamma_F = W^* (\Delta T)^2 (\Delta \alpha)^2 E_{ox} (1 - \nu_{ox})$$

□ Route II: Buckling.

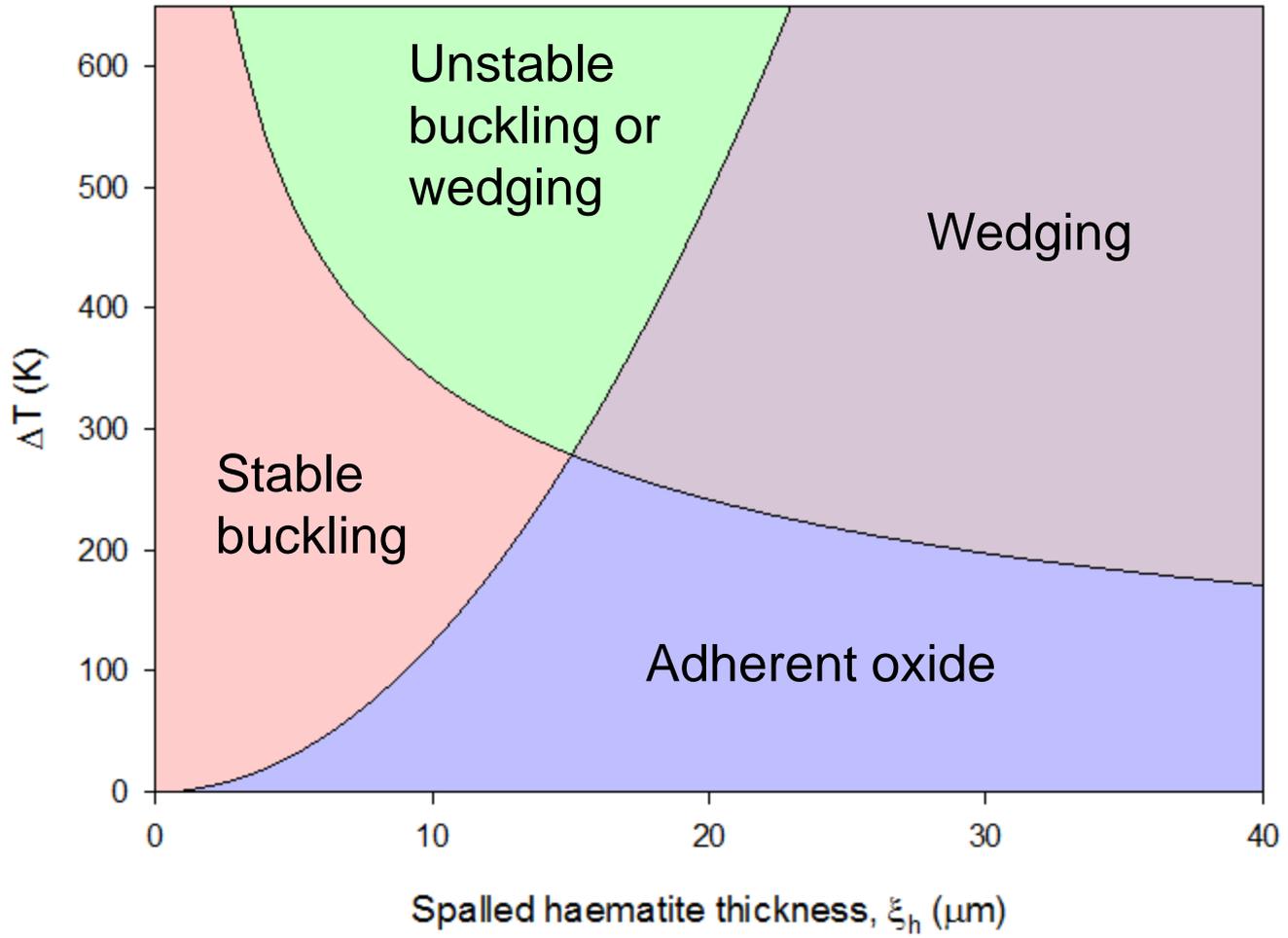
$$\Delta T_b = \frac{1.22}{(\alpha_m - \alpha_{ox}) (1 - \nu_{ox}^2)} \left(\frac{\xi}{R} \right)^2$$

E_{ox} = Young's modulus, α_m and α_{ox} = thermal expansion coefficients of metal and oxide, ν_{ox} = Poisson's ratio.

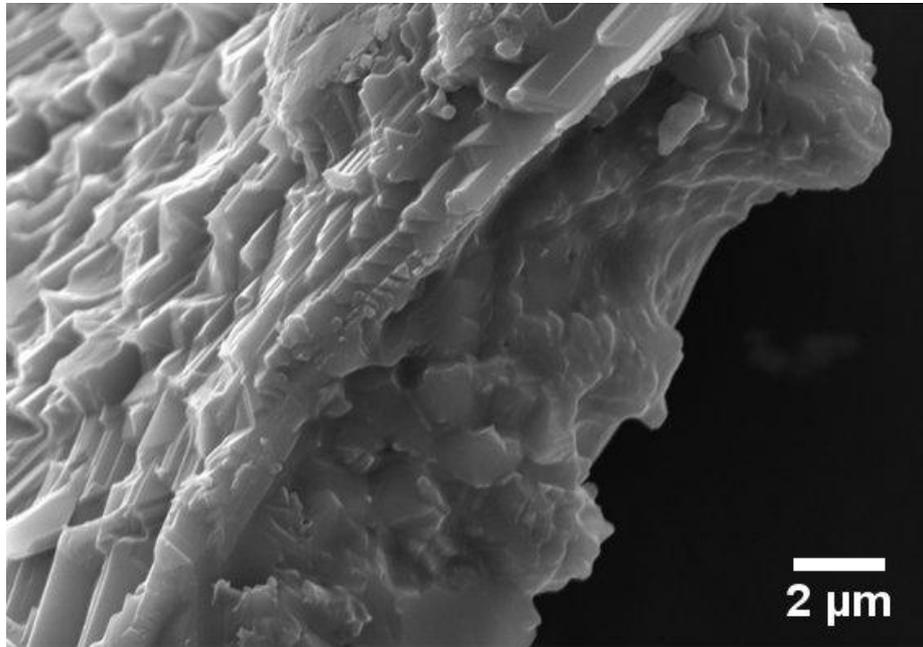
ΔT = oxidation temperature - event temperature, R = radius of delamination site, W^* = strain energy, γ_F = fracture energy, ξ = oxide thickness.



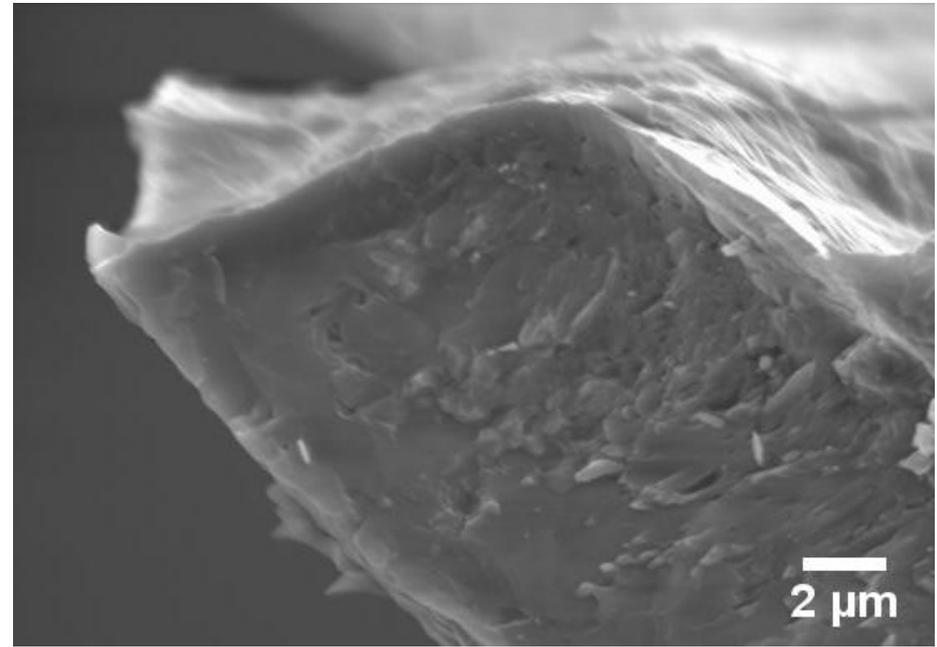
Spallation Map



Imaging Spall Particles



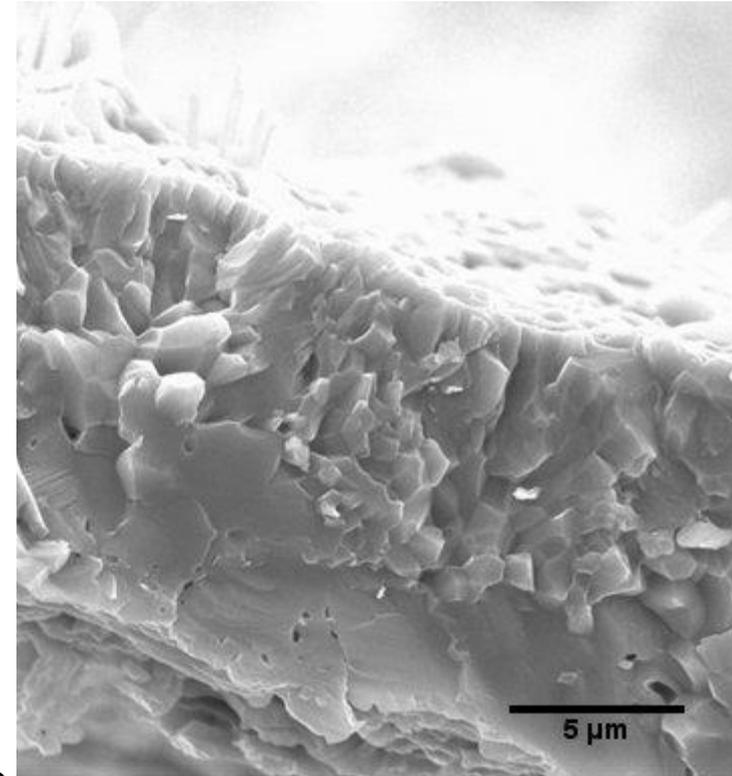
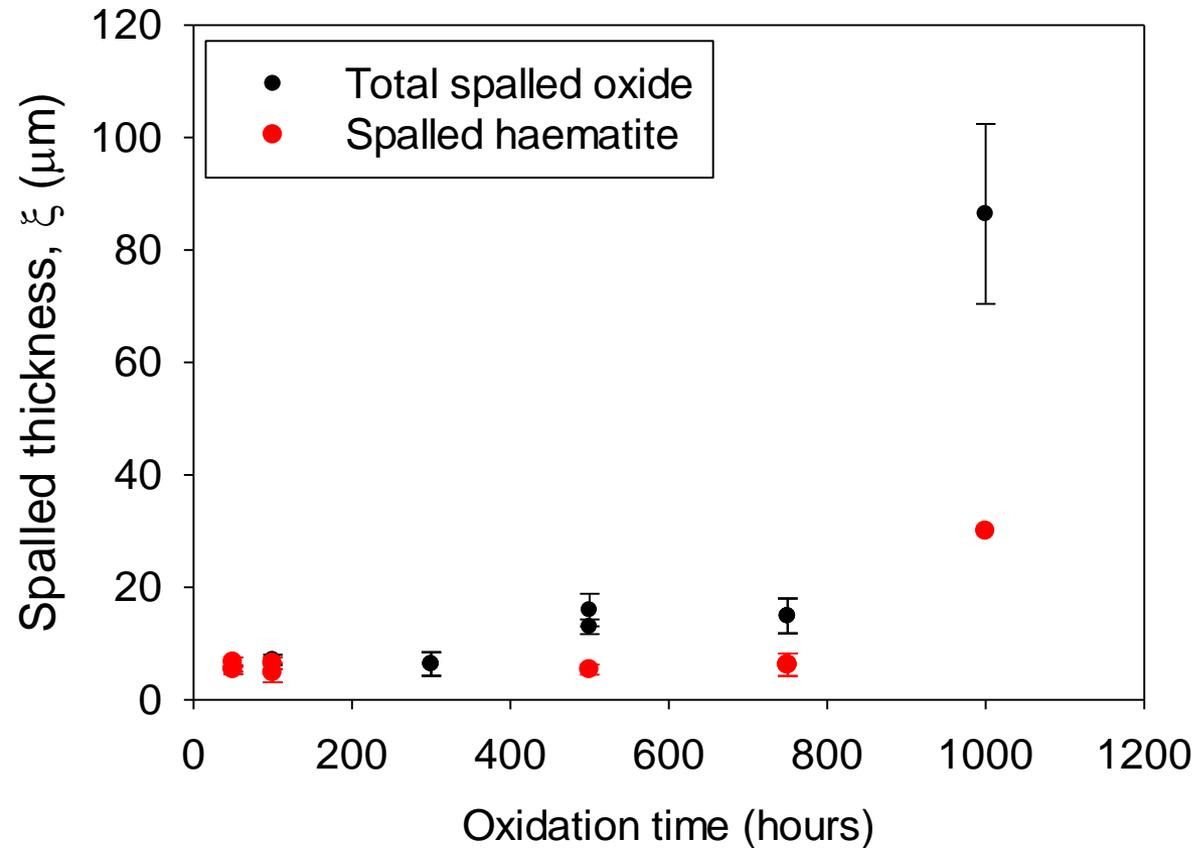
100 hours, 923 K



500 hours, 923 K



Spalled Oxide Thickness, ξ



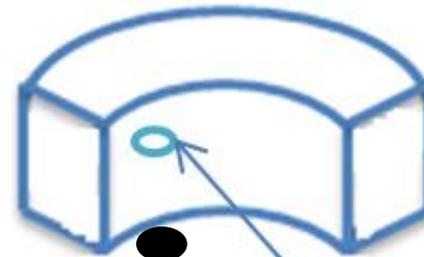
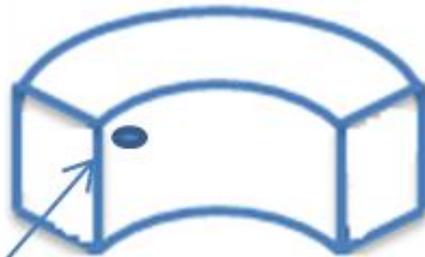
R and ΔT Measurements

- ❑ Synchronised IR and video camera.
- ❑ Able to detect delamination and spallation sites.
- ❑ Provide localised ΔT and R values.



The Measurements

During Cooling.



Initial local delamination:

- Detectable with IR camera
- T_D = temperature at which delamination occurs.

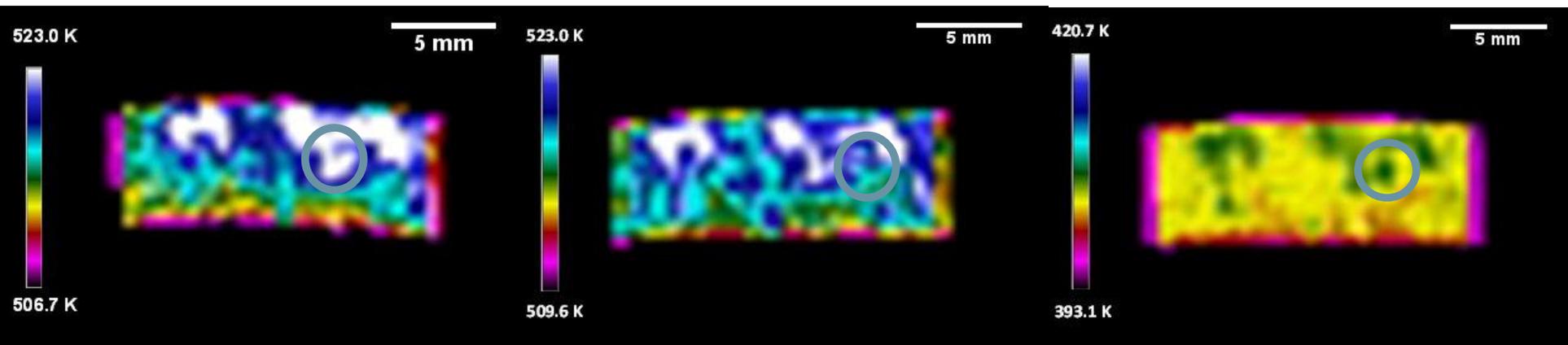
Complete local spallation:

- Detectable with video and IR camera
- T_{SP} = temperature at which spallation occurs.



IR Camera Results

- Images shown are the same sample on cooling after 100 hours in steam at 923 K.



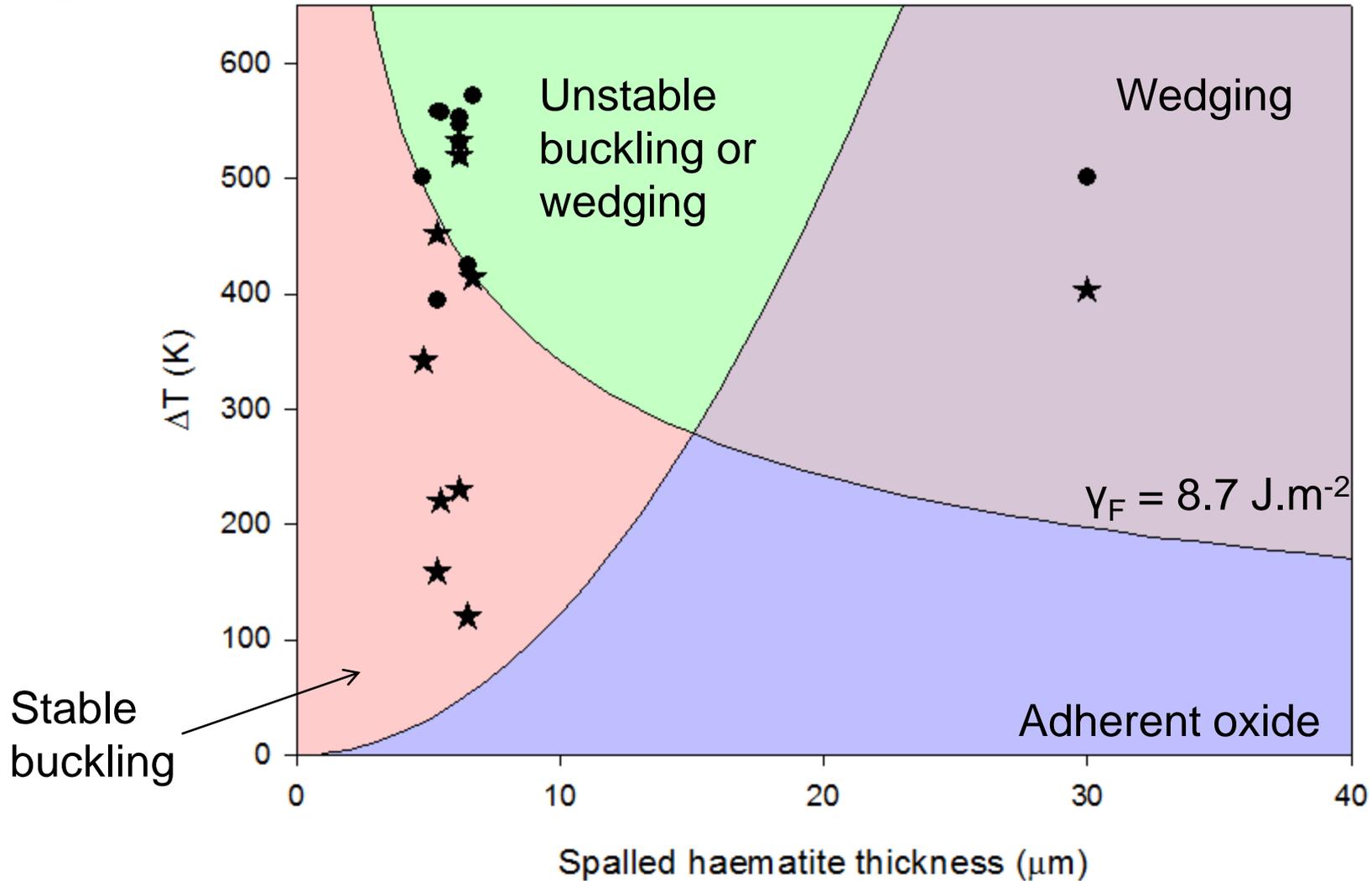
Before delamination.
 $\Delta T = 398.6$

- At time of delamination.
- $\Delta T_D = 405.5$

- At time of spallation.
- $\Delta T_{SP} = 508.9$



Spallation Map



Conclusions

- A unique combination of infrared and video camera techniques has been used to obtain critical data to input into the spallation model.
- Delamination and formation of buckles at oxide interfaces occurs – illustrated by IR camera.
- Spalled oxide thickness measurements have been obtained using high resolution scanning electron microscopy.
- An approach using fracture energy of the oxide has been shown to predict the onset of spallation.



Future Work

- Planned publication for this section of the work.
- Ongoing analysis of the degradation occurring in the alloy due to oxidation processes.
- The effect of thermal cycling.
 - Including accelerated degradation techniques.





Acknowledgements.

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